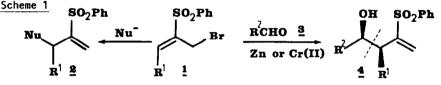
DIASTEREOSELECTIVE ADDITION OF THE 2-PHENYLSULFONYL-SUBSTITUTED ALLYLIC BROMIDES TO ALDEHYDES IN THE PRESENCE OF ZINC OR CHROMIUM (II) CHLORIDE

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Summary : The 2-phenylsulfonyl allylic bromides 1 add diasteroselectively to various aldehydes 3 in the presence of zinc or of in situ generated chromium (11) salts to give mainly the syn-hydroxy sulfones 4 in high yields. The sulfone 4b can be easily transformed into the diastereomerically pure γ -pivaloyloxy ketone 14 allowing a complete control of the stereochemistry of four adjacent chiral centers in an acyclic compound.



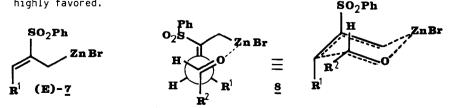
Recently we have reported that the easily available bromo-sulfones 1 of type 1 react regioselectively with nucleophiles (Nu) to produce the highly functionalized sulfones of type 2. We report now that the sulfones of type 1 are also able to react regioselectively with the aldehydes 3 in the presence of a metal 2 , or a metallic salt, to furnish the hydroxysulfones of type 4 (see scheme 1).

Thus the reaction of the bromo sulfones 1a-c (1.0eq.) with the aldehydes 3a-f (1.5eq.) in the presence of zinc³ (7eq. ; THF ; 0.5hr ; 30°-40°) or of the in situ generated chromium (II) salts^{4,5} (2.0eq. ; THF ; 1.5hr ; 25°) gives in excellent yields (83-99%) and high diastereoselectivity (generally over 90:10) the *syn* -hydroxy sulfones 4a-j (see the table and scheme 2). It is remarkable 6 that the zinc mediated reaction shows such a high diastereoselectivity (see entries 1, 2, 3 and 8 of the table). We assign this high selectivity to the presence of the phenylsulfonyl group which highly favors the (E)-configuration $^{\prime}$ for the intermediate allylic zinc compound 7. Thus a chair transition state of type 8 is favored and explains the formation of the syn -diastereoisomer 4 for the following reasons :

the steric interaction between the $PhSO_2$ - and the R^2 groups are minimized ; (i)

(ii) the R² group occupies an equatorial position in the chair transition state ; (iii) the 109° angle (Bürgi-Dunitz angle⁸) approach of the allylic reagent is sterically

highly favored.



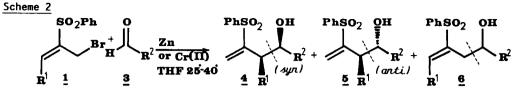


Table : Synthesis of the *ayn* hydroxy sulfones <u>4a-j</u> by the zinc, bismuth or chromium chloride mediated addition of the bromo-sulfones <u>1a-c</u> to the aldehydes <u>3a-e</u>

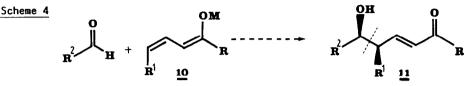
Bromo-			Aldehyde			Metal or		•	Products(%)					Yield		
Entry	sul	fone	<u>1</u>	R ¹		<u>3</u>	$ R^2$	1	letal sal	lt	4	5		6		(%) ^a
	_I		_ _		_		I			_ _		<i>.</i>			_	
1	Ι.	<u>1a</u>	Ι	Me	Ι	<u>3a</u>	Ph-	I	Zn	I	<u>4a</u> :100	<u>5a</u> :	0	<u>6a</u> :	0	95
2	Ι.	<u>1a</u>	Ι	Me	Ι	<u>3b</u>	cyclo-Hex-	Τ	Zn		<u>45</u> :100	<u>5</u> b:	0	<u>6b</u> :	0	99
3	1	<u>1a</u>	1	Me	1	<u>3c</u>	Pent~		Zn		<u>4c</u> : 88	<u>5c</u> :1	12	<u>6c</u> :	0	91
4	1	<u>1a</u>	Ι	Me		<u>3d</u>	tert-Bu-	l	Zn	I	<u>4d</u> : 83	<u>5</u> d:	0	<u>6d</u> :	17	99
5	Ι.	<u>1a</u>	1	Me	1	<u>3d</u>	tert-Bu-	1	CrCl ₂	1	<u>4d</u> : 81	<u>5</u> d:	0	<u>6d</u> :1	19]	92
6	Ι.	<u>1a</u>	Ι	Me	1	<u>3e</u>	iso-Bu-	T		I	<u>4e</u> : 99	<u>5e</u> :	1	<u>6e</u> :	0	93
7	Ι.	1a		Me	I	<u>3f</u>	(E)-CH ₃ -CH=CH-	-		I	<u>4f</u> : 90	<u>5f</u> :1	0	<u>6f</u> :	0	83
8	_	<u>1b</u>	1	Pr		<u>3a</u>	Ph-	1	Zn	I	<u>4g</u> :100	<u>5g</u> :	0	<u>6g</u> :	0	95
9		<u>1b</u>	I	Pr	1	<u>3c</u>	Pent-	1	Zn		<u>4h</u> : 67	<u>5h</u> :3	33	<u>6h</u> :	0	98
10	1	<u>1b</u>	I	Pr	Ļ	<u>3c</u>	Pent-	L		I	<u>4h</u> : 94	<u>5h</u> :	6	<u>6h</u> :	0	93
11	1	<u>1b</u>	1	Pr	1	<u>3c</u>	iso-Bu-	I		I	<u>4i</u> : 96	<u>5i</u> :	4	<u>6i</u> :	0	95
12	1	1c	i	so-B	u	<u>3c</u>	Pent-	1	CrCl ₂	1	<u>4j</u> : 90	<u>5j</u> :1	0	<u>6j</u> :	0	89
13	1	<u>1b</u>	ł	Pr	1	<u>3c</u>	Pent-	I	Bi	Ι	<u>4h</u> : 0	<u>5h</u> :5	50	<u>6h</u> :5	50	37
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a/ The given yields are those of flash chromatographically purified materials (in all cases the isomers of type <u>4</u>, <u>5</u> and <u>6</u> could be separated). The reactions are performed on a 10mmol scale. All the spectroscopic datas (I.R., H-NMR and C-NMR) are compatible with the structure shown.

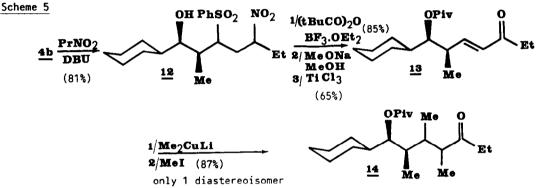
Similar discussions can be made for the chromium salts mediated reactions which show even better diastereoselectivities (compare entries 9 and 10 of the table). A preliminary result indicated that bismuth⁹ leads, although in low yield, only to the *anti*- diastereo isomer of type 5 (see entry 13). The regioisomer of type 6 is only obtained if a bulky aldehyde or if bismuth are used (see entries 4, 5 and 13). The relative stereochemistry assigned to our major isomer (of type 4) is confirmed by X-ray structures¹⁰ of compounds 4d and 9a. The functionalized tetrahydrofurans 9a and 9b are each obtained in high yields (scheme 3) as a sole isomer by a 5-Endo-Trig ring closure¹¹ respectively of the sulfones 4a and 4c.



The hydroxy sulfones of type $\underline{4}$ are very useful intermediates and allow the synthesis of products of type $\underline{11}$ resulting from a formal diastereoselective and regiospecific γ -hydroxyalkylation¹² of a dienolate of type $\underline{10}$ (see scheme 4). Thus, the addition of an excess of nitropropane to the sulfone $\underline{4b}$ in the presence of DBU¹³ gives the hydroxy



nitrosulfone <u>12</u> in 81% yield. Transformation into the pivaloyloxy derivative $((tBuCO)_2 0 excess; BF_3.0Et_2 cat.¹⁴; 1 hr; 60°; 85%) followed by a Nef reaction¹⁵ (MeONa(1.5eq.); MeOH; then TiCl_3(4eq.); NH_4OAc(24eq.); 2hr; 25°) gives, after a spontaneous elimination of PhSO₂H, the diastereoisomerically pure <math>\gamma$ -pivaloyloxy enone <u>13</u> in 65% yield.



Addition of Me₂CuLi (1.5eq. ; ether ; -15° ; 0.75 hr) followed by the addition of an excess of iodomethane (10eq. CH_3I ; HMPT/THF ; 0.25 hr ; 25°) leads to <u>only one diastereo</u>-isomer¹⁶ of the ketone 14 in 87% yield.

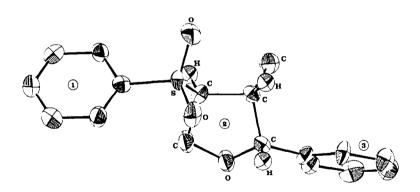
The scope of this method, which allows a complete control of the stereochemistry of four adjacent chiral centers in an acyclic compound $(\underline{1a} + \underline{14})$ as well as the determination of the relative stereochemistry of all four centers in ketone $\underline{14}$ are currently investigated in our laboratory.

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References and Notes :

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- 3. In a typical reaction, 606mg (2mmol) of the sulfone 1b in 4ml THF are slowly added (within 15min) to a THF solution of PhCHO (318mg ; 3mmol ; 2ml THF) and activated zinc (1g; 15mmol; for the zinc activation see ref. 2d) at 30° -40° in a flask immersed in a common laboratory ultrasonic cleaner (48kHz; 30w). After 0.5hr, the reaction mixture is cooled to 0° and quenched with an NH₂Cl solution. After the usual work-up and a flash-chromatography, the pure hydroxy-sulfone 4g is isolated (630mg; 95%).
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 5. In a typical reaction, a THF solution of LiAlH, (76mg; 2mmol) is added to a suspension of CrCl, (630mg; 4mmol; 5ml THF) at -15°. The resulting dark solution is stirred 15min at 20° and 606mg (2mmol) of the bromo sulfone 1b in 5ml of THF and then a THF solution of hexanal (336mg ; 3mmol ; 5ml THF) are successively added. After 1.5hr at 25°, the reaction mixture is cooled to 0° and quenched with an $\rm NH_{A}Cl$ solution. After the usual work-up and a flash-chromatography, two pure products are isolated : 570mg of the syn -hydroxy sulfone 4h and 35mg of the anti -hydroxy sulfone 5h (total yield : 93%).
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